

Enhancing Soil Remediation: Phytoremediation and Plant- Microbe Interactions for Heavy Metal Pollution

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Abstract- Heavy metal contamination in soil poses a significant threat to human health and environmental well-being. This paper explores phytoremediation as a promising, cost-effective, and eco-friendly approach for remediating such soils. We examine the nature and properties of heavy metal contaminated soils, highlighting the harmful effects on plant growth and the need for remediation. This paper focuses on two key phytoremediation techniques: phytoextraction and phytostabilization. Phytoextraction utilizes hyperaccumulating plants to absorb and accumulate metals in their above-ground parts, facilitating removal from the soil. Phytostabilization, on the other hand, immobilizes metals within the soil using plants with high metal tolerance, minimizing their mobility and preventing further environmental spread. We delve into the mechanisms employed by plants to detoxify metals, including plasma membrane barriers, antioxidant systems, intracellular chelation, and compartmentalization. While acknowledging the advantages of phytoremediation like soil conservation, cost-effectiveness, and versatility, the limitations are also discussed, such as slow process, root depth limitations, and dependence on plant age and biomass. Furthermore, the potential of plant-microbe partnerships for enhancing phytoremediation is explored. Plant growth-promoting rhizobacteria (PGPR) are highlighted for their ability to promote plant growth and tolerance to stress, ultimately improving metal uptake and stabilization.

Keywords- Phytoremediation, heavy metals, soil contamination, plant-microbe interaction, plant growth promoting rhizobacteria.

I. INTRODUCTION

Heavy metals, while naturally occurring in soil, can be further concentrated through geological processes or human activities. Common human activities contributing to increased heavy metal concentration in soil include mining, smelting, burning fossil fuels,

agricultural practices such as pesticide and fertilizer use, and industrial processes like metal production. Additionally, sewage sludge and municipal waste disposal can exacerbate soil contamination. The presence of heavy metals in soil has been observed to adversely affect plant growth and development due to alterations in their physiological and biochemical processes. Prolonged exposure to contaminated soil can result in reduced crop yields and, consequently, food insecurity. Various methods exist to remediate heavy metal-contaminated soils, broadly categorized into physical, chemical, and biological approaches. Traditional physical and chemical methods, such as encapsulation, solidification, stabilization, electrokinetics, vitrification, vapor extraction, soil washing, and flushing, often prove costly and may render the soil unsuitable for plant growth. In contrast, biological approaches, particularly bioremediation, are considered more environmentally friendly and cost-effective.

Bioremediation promotes the growth and reestablishment of plants on polluted soils by harnessing natural processes, such as the degradation of contaminants by microorganisms or the uptake of contaminants by plants. This approach is deemed economical compared to other remediation techniques and may even enhance the soil's fertility.

II. HEAVY METALS: AN ALARMING THREAT TO SOIL AND ENVIRONMENT

Heavy metal (HM) pollution constitutes a significant environmental apprehension, stemming from various anthropogenic activities such as industrial production, mining operations, and the utilization of metal-containing compounds in both domestic and agricultural spheres. Heavy metals encompass a spectrum of elements endowed with metallic

attributes, spanning transition metals, metalloids, lanthanides, and actinides. Their intrinsic toxicity, even at low exposure levels, renders them a formidable peril to living organisms. The deleterious ramifications of HMs on soil physicochemical, biological, and biochemical facets are extensively documented, prompting global concern. Their accumulation in soil and the wider environment elicits multifarious risks, encompassing ecosystem integrity, human health, food chain security, food quality assurance, and agricultural land utility. This contamination engenders perilous ramifications as heavy metals infiltrate other environmental compartments, including groundwater reservoirs, river systems, and agricultural produce, thus posing a tangible threat to human well-being.

Studies evince that heavy metals exceeding permissible thresholds precipitate a decline in water quality, rendering it unsuitable for both potable consumption and agricultural irrigation. The degree of toxicity associated with each metal is contingent upon the duration of exposure and the absorbed dosage. The enduring presence of heavy metals within soil matrices facilitates their uptake by plant tissues, consequently permeating the biosphere and imperiling human health. Hence, the imperative to avert soil contamination by heavy metals assumes paramount importance to mitigate detrimental impacts on crop yield, disrupt the integrity of the food chain, and safeguard human health. While recovery from heavy metal pollution in soil is plausible, the efficacy of remedial methodologies hinges upon the specific metals present and necessitates tailored intervention strategies.

III. IMPACT OF SOIL CONTAMINATION AND NEED FOR REMEDIATION

The rapid rise of urbanization and industrialization globally has exposed soil to an alarming degree of heavy metal (HM) contamination. This has sparked growing awareness of the link between environmental pollution and public health. Therefore, a crucial objective of any soil remediation strategy is to restore contaminated sites to a pre-pollution state, safeguarding human well-being and ensuring a sustainable environment for future generations.

Remediation of HM-contaminated soil falls under stringent regulations, often dictated by human health

assessments and ecological risk evaluations. Leaving such sites untreated poses severe long-term consequences for the environment. The choice of remediation technique hinges on factors like site specifics, contamination level, contaminant type, and intended future use of the land.

IV. REMEDIATION TECHNIQUES: TRADITIONAL VS. BIOTECHNOLOGICAL

Traditional approach

Current approaches to soil remediation typically follow two principles: complete contaminant removal or transformation into less harmful forms through engineering technologies. Decontamination methods are categorized as in situ or ex situ. Numerous physical, chemical, and biological techniques have been explored, evaluated, and implemented for HM remediation in contaminated areas. Physicochemical methods are widely used, but they suffer from drawbacks like high cost, lack of specificity, and potential damage to soil quality. This has led scientists to explore new, innovative methods with greater promise.

Biotechnological Remediation

Biotechnological methods represent an emerging and promising technology for HM-contaminated soil remediation. They offer reliable, cost-effective, and feasible alternatives compared to traditional techniques. This review focuses on two key biotechnological approaches: phytoremediation and microbial remediation. Bioremediation, in general, employs living organisms to remove and neutralize contaminants. While traditionally dominated by bacterial use, the scope has expanded to encompass a wider range of life forms, including plants (phytoremediation), fungi (mycoremediation), and even animals (zooremediation). These organisms have developed various depolluting mechanisms like biosorption, bioaccumulation, biotransformation, and biomineralization to survive in HM-contaminated soils, enabling both ex situ and in situ bioremediation. Ultimately, they render pollutants less toxic, immobilize them, or extract them. Bioremediation methods boast several advantages over traditional techniques, including cost-effectiveness and minimal environmental impact. They are broadly classified into

plant-based (phytoremediation) and microorganism-based approaches.

V. PHYTOREMEDIATION: A SUSTAINABLE APPROACH

Phytoremediation presents a highly auspicious, solar-powered, cost-effective, and ecologically sound approach to remediating soil. Its inception dates back to 1983, heralding a pioneering era in environmental restoration. Also referred to as agro-remediation, botano-remediation, or green remediation, this innovative technology harnesses the inherent capabilities of plants to extract, mitigate, transform, or immobilize contaminants, spanning both organic and inorganic pollutants present in soil, sediment, and groundwater. Certain botanical species exhibit a remarkable propensity to accumulate significant quantities of metals within their foliage and other tissues. Phytoremediation demonstrates notable efficacy, particularly in addressing soils afflicted with low to moderate levels of metal contamination. Moreover, it can synergistically complement conventional restoration methodologies to augment contaminant removal endeavors. Plants serve as active agents in concentrating and/or degrading pollutants, while their intricate root systems play a pivotal role in mitigating soil erosion and constraining the dispersal of contaminants. Furthermore, they foster a conducive physicochemical milieu within the rhizosphere, fostering the proliferation of beneficial microflora, thereby augmenting pollutant detoxification processes within the soil matrix. In comparison to conventional physicochemical remediation techniques, phytoremediation offers a plethora of advantages. Notably, it embodies environmental sustainability, cost efficiency, ease of implementation, and compatibility with diverse remediation modalities. This versatile technology is subcategorized into five distinct subclasses, namely phytoextraction, phytostabilization, phytovolatilization, phytotransformation, and phytofiltration, each offering unique mechanisms tailored to specific contaminant profiles and environmental contexts.

VI. PHYTOREMEDIATION TECHNIQUES AND FUNCTIONALITIES

Phytoremediation offers a promising, environmentally friendly approach for removing contaminants from soil and water. This technology utilizes various mechanisms employed by plants to either extract, transform, or immobilize pollutants. Here, we delve into the different phytoremediation techniques and their functionalities:

1. Phytoextraction:

- **Function:** This method involves using hyperaccumulator plants, which efficiently absorb and accumulate metals from the soil in their above-ground parts. These plants can then be harvested and processed to recover the metals, potentially for reuse (phytomining).
- **Advantages:** Effective for removing metals from contaminated soil without harming soil fertility.
- **Limitations:** Requires specific plant species, repeated harvesting cycles, and appropriate metal recovery methods.

2. Phytostabilization:

- **Function:** This strategy focuses on immobilizing contaminants within the soil using plants. Plants with high metal tolerance and minimal translocation of contaminants to their shoots are used. This prevents further spread of pollutants and protects the environment.
- **Advantages:** Reduces mobility and bioavailability of contaminants, stabilizes contaminated sites, and restores vegetation cover.
- **Limitations:** Doesn't remove contaminants from the soil but only reduces their mobility.

3. Phytovolatilization:

- **Function:** This technique involves plants taking up contaminants like mercury, selenium, or arsenic and converting them into volatile forms released into the atmosphere through transpiration.
- **Advantages:** Offers complete removal of contaminants from the soil without harvesting plants.
- **Limitations:** Requires specific plants with volatilization capabilities, potential for redeposition of volatilized contaminants, and careful monitoring of air emissions.

4. Phytotransformation:

- **Function:** This method utilizes plant enzymes to break down organic pollutants into less harmful or non-toxic forms within the plant tissues.
- **Advantages:** Effective for remediating organic contaminants, eco-friendly, and doesn't require specialized plants.
- **Limitations:** Not applicable to heavy metals, requires specific enzymes and optimal conditions for degradation, and may generate secondary waste.

5. Phytofiltration:

- **Function:** This technique utilizes plant roots, shoots, or seedlings to remove contaminants from water sources. Plants accumulate or adsorb pollutants, purifying the water through various mechanisms.
- **Advantages:** Cost-effective, eco-friendly, and suitable for treating various water contaminants.
- **Limitations:** Requires specific plant species for different contaminants, needs proper maintenance and management, and may generate biomass waste.

VII. PLANT MECHANISMS FOR DETOXIFICATION:

Plants have evolved various mechanisms to tolerate and detoxify heavy metals:

- **Plasma membrane:** Acts as a barrier, limiting metal uptake and modifying ionic efflux.
- **Antioxidant system:** Enzymes like superoxide dismutases and catalases scavenge reactive oxygen species generated by metal stress.
- **Intracellular chelation:** Specific ligands like organic acids and phytochelatins bind metals, reducing their toxicity within the cell.
- **Compartmentalization:** Metals are sequestered in vacuoles or other inactive cellular compartments to minimize damage.

Phytoremediation: Evolving Technique for Contaminant Removal

Phytoremediation offers a unique biological approach for tackling heavy metal (HM) contamination in soil. It's in situ nature, environmental friendliness, and cost-effectiveness make it an attractive solution

compared to traditional methods. Key advantages include:

- **Soil Conservation:** Vegetation cover protects topsoil and structure, promoting soil integrity and preventing erosion.
- **Economic Choice:** Requires minimal excavation, equipment, and labor, lowering costs.
- **Eco-Friendly:** Minimizes environmental impact and aids ecosystem restoration, garnering public acceptance.
- **Versatility:** Applicable to a wide range of organic and inorganic soil and water contaminants.
- **Aesthetic Enhancement:** Improves and maintains landscape integrity by establishing plant cover.
- **Additional Benefits:** May enhance soil fertility, reduce residue generation, and minimize contaminant spread.

VIII. LIMITATIONS OF PHYTOREMEDIATION

- **Slow Process:** Remediation can take months to years, demanding long-term commitment.
- **Root Depth Limitation:** The extent of remediation is constrained by the root system's depth, reaching only accessible areas.
- **Biomass Dependence:** Low-biomass plants are less effective in extracting and stabilizing contaminants.
- **Environmental Factors:** Climatic conditions, soil properties, and biotic stressors can hinder plant growth and efficacy.
- **Food Chain Risk:** Absorbed contaminants in harvested biomass pose potential harm to animals and humans if not managed and disposed of properly.
- **Plant Age Influence:** Younger plants accumulate more metals than older ones, necessitating consistent population maintenance.
- **HM Level Impact:** High HM levels can induce phytotoxicity, restricting applicability.
- **Native Plant Preference:** Using native plants minimizes biodiversity disruption.
- **Multifactorial Dependence:** Bioavailability, soil properties, metal speciation, and plant species significantly influence success.

IX. EXPLOIT PLANT-MICROBE PARTNERSHIPS FOR ENHANCED REMEDIATION

The interplay between plants and microorganisms represents a promising avenue for heavy metal (HM) remediation. This symbiotic relationship, particularly within the rhizosphere, empowers plants to better withstand contaminants and elevate the efficiency of remediation efforts. Key microbial players in this symbiosis include rhizobacteria, mycorrhizae, and yeast, which can be deliberately introduced to enhance plant growth, tolerance, and overall performance. Plant Growth-Promoting Rhizobacteria (PGPR) stand out among these microorganisms, holding immense potential for bolstering phytoremediation endeavors. PGPR exert beneficial effects on plants by promoting growth and fitness, augmenting nutrient uptake, fortifying plants against diverse stresses, and enhancing soil health. These benefits are realized through a variety of mechanisms:

- Nitrogen Fixation: PGPR contribute to atmospheric nitrogen fixation, thereby enriching the soil with this crucial plant nutrient.
- Phosphorus and Potassium Solubilization: PGPR solubilize phosphorus and potassium, thus increasing the availability of these essential nutrients to plants.
- Phytohormone Production: PGPR produce phytohormones that stimulate cell division, growth, and root development in plants.
- ACC Deaminase Activity: PGPR possess ACC deaminase activity, which reduces ethylene levels in plants, thereby alleviating stress and promoting root growth.
- Phosphate Solubilization: PGPR aid plants in withstanding heavy metal stress by solubilizing phosphate in the soil.

Through their role in promoting plant health and resilience, PGPR indirectly contribute to phytoremediation through various mechanisms:

- Metal Solubilization: PGPR facilitate the uptake of metals by plants by solubilizing them in the soil.
- Siderophore Production: PGPR produce siderophores, which enhance iron acquisition in plants, thereby influencing heavy metal uptake.
- Biosurfactant Production: PGPR produce biosurfactants that increase the bioavailability of hydrophobic contaminants in the soil.

- Redox Transformations: PGPR catalyze redox transformations that convert metals into less bioavailable forms.

- Biosorption: PGPR bind metals directly to their cells, thereby reducing metal bioavailability in the soil and sediments.

These diverse mechanisms collectively enhance the efficacy of phytoremediation processes by reducing the bioavailability of metals in the soil and sediments. For example, studies have demonstrated that inoculating alfalfa with a consortium of bacteria improves seed germination, plant growth, and tolerance to heavy metal stress, ultimately resulting in reduced heavy metal accumulation and enhanced phytostabilization efficiency.

X. CONCLUSION

Phytoextraction, also referred to as phytoaccumulation or phytoabsorption, represents an in situ phytoremediation technique that exploits hyperaccumulating plants to uptake metals from the soil via their root systems and sequester them in their above-ground biomass, predominantly in the stems and leaves. This method is widely recognized for its efficacy in eliminating heavy metals (HMs) from contaminated soils while preserving soil fertility. Following harvest, metals accumulated within the plant tissues can be recovered and repurposed for economic gain through processes such as phytomining, which involves thermal, chemical, or microbiological treatments. Notably, hyperaccumulative plants, predominantly belonging to the Brassicaceae family, demonstrate a remarkable capacity for HM accumulation, exemplified by species such as *Brassica juncea*, *Thlaspi caerulescens*, *Pteris vittata*, *Haumaniastrum robertii*, *Aeolanthus biformifolius*, *Astragalus bisulcatus*, and *Arabis paniculata*. In contrast, phytostabilization, or phytoimmobilization, has emerged as an alternative remediation strategy centered on establishing vegetative cover on contaminated sites. This method entails utilizing plants to immobilize contaminants within the soil matrix through mechanisms such as root uptake, adsorption onto root surfaces, or precipitation within the rhizosphere. By doing so, phytostabilization effectively reduces the solubility, bioavailability, and potential harm of HMs in the environment. While it does not serve as a

comprehensive clean-up technique, phytostabilization mitigates contaminant mobility, thus minimizing the risk of further environmental degradation. To execute phytostabilization successfully, plants must exhibit tolerance to elevated levels of toxic contaminants, possess extensive root systems, and demonstrate limited translocation of contaminants from roots to shoots. Utilization of metal-tolerant plant species facilitates the restoration of vegetative cover on contaminated sites, thereby mitigating the dispersion of toxic substances via mechanisms such as wind dispersal, surface runoff, and leaching into groundwater. Consequently, phytostabilization contributes to the enhanced integration of contaminated sites into the broader landscape.

Both phytoextraction and phytostabilization offer environmentally sustainable and efficient approaches for the remediation and immobilization of contaminants, particularly heavy metals, in soil environments. The successful implementation of these phytoremediation techniques hinges upon careful selection of suitable plant species, consideration of site-specific characteristics, and assessment of the nature and extent of contamination. Continued research and development efforts are imperative to further refine and optimize these phytoremediation methodologies, thereby bolstering their effectiveness and applicability in addressing environmental contamination challenges.

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