Design, Molecular Docking, Synthesis and Biological Study of Pyrazole Containing Hydrazinyl Pyrimidine Derivatives with Potent Antitubercular Activity

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Abstract—A series of novel pyrazole-containing hydrazinyl pyrimidine derivatives were designed and subjected to molecular docking studies to investigate the binding interactions of the designed compounds with Protein (PDB ID: 5V3Y) involved in the survival of Mycobacterium tuberculosis. The Compounds 2a-2f shows favourable binding affinities and provided insight into the mechanisms of action. Those good binding energy compounds were synthesized from substituted acetophenone and substituted benzaldehyde to get the pyrazole-containing hydrazinyl pyrimidine derivatives and characterized by analytical IR, ¹HNMR and ESI-Mass spectrometry. Later, the synthesized compounds were evaluated for in-vitro anti-tubercular activity against Mycobacterium tuberculosis for MTCC 300 strains, where the Compound 2a, 2b and 2d revealed the good antitubercular activity when compared with Rifampicin as the standard drug, suggests that these pyrazole-pyrimidine derivatives hold potential as effective candidates for the development of new antitubercular agents.

Index Terms—Pyrazole, Hydrazinyl Pyrimidine, Molecular docking, Antitubercular Activity.

I. INTRODUCTION

Tuberculosis (TB), caused by *Mycobacterium tuberculosis*, remains a major global health challenge claiming millions of lives each year ^[1]. Despite the availability of treatment, the emergence of multidrugresistant and extensively drug-resistant strains has highlighted the need for novel and more effective antitubercular agents. In response to this urgent need, the design of new compounds targeting key biological pathways in *M. tuberculosis* has become a priority in drug discovery ^[2].

Pyrazole-containing compounds have garnered attention for their diverse pharmacological activities including antimicrobial and antitubercular effects. The incorporation of hydrazinyl pyrimidine moieties into such structures is particularly promising as pyrimidine derivatives are known for their role in inhibiting enzymes critical to the survival of mycobacteria. By combining these two bioactive scaffolds, we aim to develop novel molecules with enhanced antitubercular potency [3].

This study focuses on the synthesis of a series of pyrazole-containing hydrazinyl pyrimidine derivatives, followed by biological testing to evaluate their antitubercular activity against *M. tuberculosis* ^[4]. In addition, molecular docking studies were conducted to predict the binding affinity of these compounds to key target enzymes, providing valuable insight into their mechanism of action. The findings of this research could lead to the identification of promising drug candidates, contributing to the development of more effective treatments for tuberculosis, especially against resistant strains ^[6].

II. MATERIALS AND METHODOLOGY

A. Synthesis:

Step 1: Synthesis of N-acetyl pyrazole (1a, 1b, 1c, 1d, 1e, 1f): A solution of substituted acetophenones (0.01mol) in minimum quantity of ethanol (10 mL) was added to a benzaldehyde (0.01mol) in ethanol (10 mL) stir the mixture. To that add hydrazine hydrate (0.01M) with KOH solution. The reaction mixture was refluxed for 8hours, cool, diluted with water and acidified with conc. HCl. The product was filtered, dried and crystallized from ethanol.

Fig no 1: Scheme for the synthesis of pyrazole contains hydrazinyl pyrimidine derivatives

Step-2: Synthesis of novel pyrazole contains hydrazinyl pyrimidine derivatives (2a, 2b, 2c, 2d, 2e and 2f): To take 0.005mol of compound (Pyrazole derivatives), 5.1ml of acetic anhydride was added and warm hydrazine hydrate (0.01mol) solution, 2-amino pyrimidine (0.01mol) in alcohol (30ml) and NaOH (5ml, 0.01mol) solution was added and refluxed for 6h in the presence of glacial acetic acid. The progress of the reaction was monitored by TLC. The reaction mixture was cooled to room temperature and kept in refrigerator for overnight and extracted the product by using ethyl acetate. This was filtered off and recrystallized from ethanol to give crystalline solid.

III. MOLECULAR DOCKING

A. Dataset ligands and Ligand Optimization

The 2D structures of 6 compounds were generated from the ACD/Chemsketch Software. The generated ligands cleaned and performed 3D optimization then saved in the MDL Molfile format. The ligands were then converted to a PDBQT file format using the Open Babel chemistry toolbox ^[6].

B. Molecular Docking Studies

The three-dimensional structure of Pokeweed Antiviral Protein (PDB ID: 5V3Y) was downloaded from Brook heaven Protein Data Bank (https://www.rcsb.org) and saved as a Brookhaven

protein data bank file and the structure was optimized by deleting unbound water molecules which are over 1 Å, adding hydrogen atoms to satisfy the valences, adding missing amino acids to stabilize side chains and energy of the whole structure was minimized using AUTODOCK suite of MGL Tools ^[7].

Auto dock Vina was used for molecular docking studies. A grid was generated around the cocrystallized ligand. The co-ordinates (x=32.63, y=26.25, z=16.59) were generated with the help of MGL Tools & Pharmit: interactive exploration of chemical space (http://pharmit.csb.pitt.edu/). Prepared pdbqt files for both target & ligands. Created in house batch file of ligands & amp; target and docking performed in the absence of water molecules for all 15 molecules. The molecules were analyzed after docking and visualized in the discovery studio for the interactions with the active site amino acids [8].

Binding interactions and efficiency of the binding were calculated in terms of dock Score, which is a combination of hydrophilic, hydrophobic, metal binding groups, Van der Waals energy, freezing rotatable bonds and polar interactions with receptor [9,10]

C. Characterisation:

Melting point of the synthesized compounds was determined by an open-end capillary tube method using electrically heated melting point apparatus. The

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respective values were expressed in oC and were uncorrected. Reaction progress and compounds purity was ascertained by thin layer chromatography (TLC). The structures of the synthesized compounds were elucidated by Fourier Transform IR spectrometer (Thermo Nicolet Nexus 670) in the range of 400-4000 cm⁻¹ using KBr pellets and values are reported in cm⁻¹ and the spectra were interpreted. 1 H-NMR spectra were recorded on Bruker-Topspin NMR spectrometer using DMSO-d6 and chemical shift (δ) are reported in parts per million down field from internal reference Tetramethylsilane (TMS). Mass spectra were recorded on Shimadzu by LC-MS-8030 mass spectrometer and the spectra were interpreted.

D. Antitubercular Activity:

The M. tuberculosis (MTCC 300) strain was used in present study for the assessment antimycobacterial activity. The strain used for the study was procured from Microbial Type Culture and Gene Bank, Institute of Microbial Technology, Chandigarh (PB), India. The mentioned strain was sub-cultured and persevered as per the earlier depicted method on the Lowenstein Jensen medium. The Agar diffusion method was used to assess the sensitivity of M. tuberculosis strain against the synthesized compounds. Different stock solutions such as 0.1, 0.5, and 1.0 mg/mL of all compounds were prepared in dimethyl sulfoxide (DMSO). A sterile corn borer of 9mm diameter was further employed to prepare holes into the Middlebrook 7H9 agar, already inoculum seeded and solidified. Firstly, the wells were labelled

appropriately according to the compounds, and afterward a volume of 40 μ L of each compound was added by using a sterile pipette. The test was executed in triplicates. To achieve the sample pre-diffusion, the plates were stored in the refrigerator and further incubated at room temperature for 48 h. After the incubation, the growth of the mentioned strain was detected and the diameter of the inhibition zone was measured. Rifampicin was used as a positive control for the mentioned experiment [11].

IV. RESULT AND DISCUSSION

A. Molecular Docking:

Molecular docking studies were performed in order to find the possible protein ligand interactions of the dataset ligands. The potential active site amino acids of 5V3Y complex were predicted using CASTp. The target protein and inhibitors were geometrically optimized. All the 6 compounds were docked against active site of target protein using AUTODOCK VINA. Additionally, these also assisted in identifying the conformational changes of the ligand in the protein environment. About 100 different protein-ligand complex conformations for each docked complex were generated through AUTODOCK suite of MGL Tools, the confirmation with lowest binding energy was displayed as the best binding energy. Binding energy of the dataset ligands were shown in Table 1 along with the interaction amino acids and number of amino acids.

Compound	Structure	Binding	No of	Hydrogen Bonding	Van der Waal's
name		energy	Hydrogen	Amino acids	Interaction
			bonding	residues	
2a		-10.2	1	TYR;1582	VAL;1611,ALA;1586
	ÇH ₃				,
	NH N				TRP;1579,ILE;1594,
					ALA;1583,ILE;1597,
					ARG;1581,MET;166
					9
2b	H₃C	-9.8	1	TYR;1582	PHE;1590,ALA;1586
					,TRP;1579,ILE;1594,
					ALA;1583,ILE;1597,
	CH₃				MET;1669,
	N N NH N				ARG;1581
)—N N				

2c	CI	-9.6	1	TYR;1582	TYR;1637,PHE;1590
20		-9.0	1	11K,1362	
					,ALA;1586,TRP;157
	ÇH ₃				9,ILE;1594,ALA;158
	N NH N				3,
	>=\ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \				ILE;1597,ARG;1581,
					MET;1669
2d	F	-10.1	1	TYR;1582	VAL;1611,PHE;1590,
					TRP;1579,ALA;1586
					,ILE;1594,ALA;1583
	CH ₃				,
	N N NH N				ILE;1597,ARG;1581
	N N				
2e	H ₃ C_0	-9.6	1	TYR;1582	ARG;1634,GLN;163
					3,ALA;1586,TRP;15
					79,ILE;1594,ALA;15
	CH ₃				83,
	N NH N				ILE;1597,ARG;1581,
					MET;1669
2f	H ₃ C N CH ₃	-9.7	1	TYR;1582	ARG;1634,LEU;161
					5,ALA;1586,ILE;159
					4,ALA;1583,ILE;159
	ÇH₃				7,
	N NH N				TRP;1579,
					ARG;1581,
					MET;1669

Table no 1: Molecular Docking results with β -Tubulin.

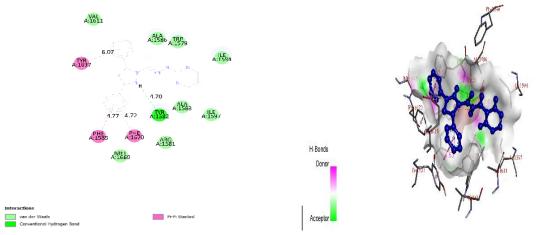


Fig no 2:2D and 3D Structures of Compound 2a with PDB id: 5v3y

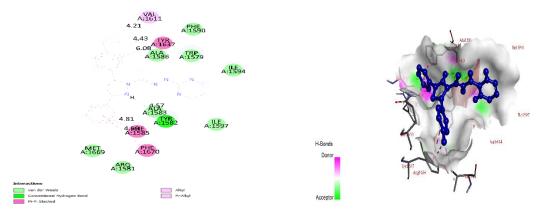


Fig no 3:2D and 3D Structures of Compound 2b with PDB id: 5v3y

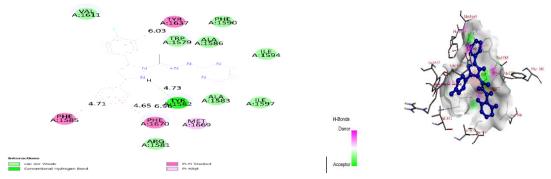


Fig no 4:2D and 3D Structures of Compound 2d with PDB id: 5v3y

B. Characterization

Compound 2a: IR(vcm-1): 3325 (N-H *Str* in amine group), 3022 (C-H *Str*; aromatic), 2973, 2897 (C-H *Str* in aliphatic), 1646 (-C=N, *Str in* immine group), 1613 (-C=N, *Str* in rings), 1405 (CH=CH *Str* in aromatic ring), 1309 (C=C, *Str* in aromatic ring).

¹H-NMR(DMSO) δ ppm: 9.42992 (s, 1H, amine), 8.8090-8.5691 (d, 2H, aromatic H), 8.2210-8.2062 (d, 2H, aromatic -H), 8.1492-8.1292 (d, 2H, aromatic -H), 8.0568-8.0384 (t, 3H, aromatic -H), 7.8501-7.7337 (t, 3H, aromatic -H), 7.7285-7.5435 (t, 1H, aromatic - H), 4.5827-4.5802 (dd, 2H in pyrazole ring protons), 3.1528 (t, 1H in pyrazole ring protons), 1.8338 (s, 3H, -CNCH3).

Mass (ESI-MS): m/z 356.17 (M), 357.21 (M + 1, 100%).

Compound 2b:

IR(vcm-1): 3227 (N-H *Str* in amine group), 3017 (C-H *Str*; aromatic), 2915, 2873, 2741 (C-H *Str* in aliphatic), 1617 (-C=N, *Str in* immine group), 1513 (-C=N, *Str* in rings), 1430 (CH=CH *Str* in aromatic ring), 1322 (C=C, *Str* in aromatic ring).

¹H-NMR(DMSO) δ ppm: 9.4031 (s, 1H, -NH proton), 8.4787-8.3533 (d, 2H, aromatic H), 8.0122-8.0012 (d, 2H, aromatic -H), 7.9492-7.9267 (d, 2H, aromatic -H),

7.8287-7.7574 (d, 2H, aromatic -H), 7.1705-7.1463 (t, 3H, aromatic -H), 6.8819-6.8350 (t, 1H, aromatic -H), 4.3817- 4.3051 (dd, 2H in pyrazole ring protons), 3.0174 (t, 1H in pyrazole), 2.0714 (s, 3H, -CN-CH3), 1.8670 (s, 3H, Ar-CH3).

Mass (ESI-MS): m/z 370.19 (M), 371.3 (M + 1, 100%).

Compound 2c:

IR(vcm-1): 3205 (N-H *Str* in amine group), 3021 (C-H *Str*; aromatic), 2931, 2893 (C-H *Str* in aliphatic), 1616 (-C=N, *Str in* immine group), 1602 (-C=N, *Str* in rings), 1381 (CH=CH *Str* in aromatic ring), 1345 (C=C, *Str* in aromatic ring), 787 (C-Cl, *Str* in Ar-Cl ring).

¹H-NMR(DMSO) δ ppm: 9.1591 (s, 1H, -NH proton), 8.5001-8.3190 (d, 2H, aromatic H), 8.1546-8.1139 (d, 2H, aromatic -H), 8.0910-8.0142 (d, 2H, aromatic -H), 7.9400-7.9068 (d, 2H, aromatic -H), 7.8989-7.8040 (t, 3H, aromatic -H), 7.7940-7.6778 (t, 1H, aromatic -H), 4.5330-4.5078 (dd, 2H in pyrazole ring protons), 3.1726 (t, 1H in pyrazole), 2.0480 (s, 3H, Ar-CH3). Mass (ESI-MS): m/z 390.14 (M), 391.03 (M + 1, 100%), 392.54 (M + 2, 30%)

Compound 2d:

IR(vcm-1): 3384 (N-H *Str* in amine group), 3087 (C-H *Str*; aromatic), 2993, 2821 (C-H *Str* in aliphatic), 1632 (-C=N, *Str in* immine group), 1598 (-C=N, *Str* in rings), 1365 (CH=CH *Str* in aromatic ring), 1301 (C=C, *Str* in aromatic ring), 821 (C-CF, *Str* in Ar-F ring).

¹H-NMR(DMSO) δ ppm: 9.5321 (s, 1H, -NH proton), 8.3452-8.2901 (d, 2H, aromatic H), 8.1032-8.1092 (d, 2H, aromatic -H), 7.8943-7.6753 (d, 2H, aromatic -H), 7.4132-7.3512 (d, 2H, aromatic -H), 7.2094- 7.1293 (t, 3H, aromatic -H), 6.9396-6.7832 (t, 1H, aromatic - H), 4.3523-4.2031 (dd, 2H in pyrazole ring protons), 3.0432 (t, 1H in pyrazole), 2.0934 (s, 3H, Ar-CH3). Mass (ESI-MS): m/z 374.17 (M), 375.18 (M + 1, 100%), 376.03 (M + 2, 30%).

Compound 2e:

IR(vcm-1): 3239 (N-H *Str* in amine group), 3078 (C-H *Str*; aromatic), 2967, 2898, 2756 (C-H *Str* in aliphatic), 1631 (-C=N, *Str in* immine group), 1595 (-C=N, *Str* in rings), 1402 (CH=CH *Str* in aromatic ring), 1094 (C-O *Str* in Ar–OCH3), 1313 (C=C, *Str* in aromatic ring).

¹H-NMR(DMSO) δ ppm: 9.2093 (s, 1H, -NH proton), 8.2094-8.1732 (d, 2H, aromatic H), 8.1293-8.0021 (d,

2H, aromatic -H), 7.9845-7.8732 (d, 2H, aromatic -H), 7.5093-7.4321 (d, 2H, aromatic -H), 7.19943-7.1032 (t, 3H, aromatic -H), 6.9892-6.8773 (t, 1H, aromatic -H), 4.5643-4.4895 (dd, 2H in pyrazole ring protons), 3.6732-3.6003 (s, 3H, Ar-OCH3), 3.2187 (t, 1H in pyrazole), 1.9973 (s, 3H, Ar-CH3).

Mass (ESI-MS): m/z 386.19 (M), 387.02 (M + 1, 100%).

Compound 2f:

IR(vcm-1): 3342 (N-H *Str* in amine group), 3035 (C-H *Str*; aromatic), 2956, 2845, 2787 (C-H *Str* in aliphatic), 1643 (-C=N, *Str in* immine group), 1556 (-C=N, *Str* in rings), 1409 (CH=CH *Str* in aromatic ring), 1321 (C=C, *Str* in aromatic ring).

¹H-NMR(DMSO) δ ppm: 9.4320 (s, 1H, -NH proton), 8.4093-8.3823 (d, 2H, aromatic H), 8.1342-8.0932 (d, 2H, aromatic -H), 7.9874-7.7322 (d, 2H, aromatic -H), 7.8765-7.6433 (d, 2H, aromatic -H), 7.3231-7.2093 (t, 3H, aromatic -H), 7.2013-7.0322 (t, 1H, aromatic - H), 4.3945-4.3803 (dd, 2H in pyrazole ring protons), 3.3092 (t, 1H in pyrazole), 2.8791-2.9932 (s, 6H, Ar-N(CH3)2), 2.0321 (s,3H, Ar-CH3).

Mass (ESI-MS): m/z 399.22 (M), 400.32 (M + 1, 100%).

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Compound	R_1	R_2	Molecular	Molecular	Melting	% Yield	Rf Value
			Formula	Weight	Point		
				Gm/mol	(°C)		
2a	Н	Н	$C_{21}H_{30}N_6$	356.42	177-179	76	0.81
2b	Н	CH ₃	C ₂₂ H ₂₂ N ₆	370.04	243-245	80	0.67
2c	Н	Cl	$C_{21}H_{19}N_6Cl$	390.14	213-215	74	0.81
2d	Н	F	C ₂₁ H ₁₉ N ₆ F	374.17	209-211	74	0.73
2e	Н	OCH ₃	C ₂₂ H ₂₂ N ₆ O	386.19	201-203	69	0.76
2f	Н	N(CH3) ₂	$C_{23}H_{25}N$	399.13	219-221	78	0.72

Table No 2: Physical Characterisation data of the Synthesized Compounds (2a-2f)

C. Antitubercular Activity:

The results of the antimycobacterial activity of the synthesized compounds are summarized in Table 2, which clearly shows the differential sensitivity of Mycobacterial strain MTCC 300 toward the test compounds. The compounds $2b(\text{MIC-}12.5\pm0.61\mu\text{g/ml})$, $2b(\text{MIC-}9.0\pm0.56\mu\text{g/ml})$ and $2d(\text{MIC-}11.8\pm0.65\mu\text{g/ml})$ were found to be most effective growth inhibitors of this strain when Compared with the Standard drug Rifampicin.

Sl No	Compound name	$MIC(\mu g/ml)$
1	2a	12.5 ± 0.61
2	2b	9.0 ± 0.56
3	2c	7.6 ± 0.83

4	2d	11.8 ± 0.65
5	2e	7.6 ± 0.72
6	2f	8.7 ± 0.48
7	Rifampicin	0.8 ± 0.70

Table no 3: Antitubercular screening of compounds. The results are expressed as the mean values from three independent experiments \pm standard deviation

V. CONCLUSION

In conclusion, molecular docking studies, synthesis and biological evaluation of pyrazole-containing hydrazinyl pyrimidine derivatives have demonstrated their promising antitubercular activity against

Mycobacterium tuberculosis. The molecular docking studies provided valuable insights into the compounds' binding interactions with PDB ID: 5V3Y crucial to the survival of M. tuberculosis, supporting the observed Antitubercular activity. The in-vitro antitubercular activity revealed that Compound 2a, 2d and 2b of the synthesized compounds exhibit potent inhibitory effects, comparable to or exceeding the activity of standard antitubercular drug Rifampicin. These findings suggest that pyrazole-pyrimidine derivatives are promising candidates for further development as novel antitubercular agents, particularly for combating drug-resistant strains.

REFERENCES

- [1] Wang Y, Liu T X and Wang T Y. Isobavachalcone inhibits pseudorabies virus by impairing virus induced cell to cell fusion. Virol J. 2020;17(1):39-43.
- [2] Diaz Sanchez M, Diaz Garcia D, Prashar S and Gomez Ruiz S. Palladium nanoparticles supported on silica, alumina or titania: greener alternatives for suzuki– miyaura and other C–C coupling reactions. Environ Chem Lett. 2019;17(4):1585–602. https://doi.org/10.1007/s10311-019-00899-5
- [3] Faisal M, Saeed A, Hussain S, Dar P and Larik FA. Recent developments in synthetic chemistry and biological activities of pyrazole derivatives. J Chem Sci. 2019;131(8):1-30.
- [4] Motamedi A, Sattari E, Mirzaei P, Armaghan M and Bazgir A. An efficient and green synthesis of phthalide fused pyrazole and pyrimidine derivatives. Tetrahedron Lett. 2014;55(15):2366-8.
- [5] Safaei S, Mohammadpoor-Baltork I, Khosropour AR, Moghadam M, Tangestaninejad S and Mirkhani V. Copper (II) ionic liquid catalyzed cyclization— aromatization of hydrazones with dimethyl acetylenedicarboxylate: A green synthesis of fully substituted pyrazoles. New J Chem. 2013;37(7):2037-42.
- [6] Novikov FN and Chilov GG. Molecular docking: theoretical background, practical applications and perspectives. Mendeleev Commun. 2009;5(19):237-42.

- [7] Kroemer RT. Structure-based drug design: docking and scoring. Curr Protein Pept Sci. 2007;8(4):312-28.
- [8] Meng EC, Shoichet BK and Kuntz ID. Automated docking with grid-based energy evaluation. J Comput Chem. 1992;23(4):505-24.
- [9] Kitchen DB, Decornez H, Furr JR and Bajorath J. Docking and scoring in virtual screening for drug discovery: methods and applications. Nat Rev Drug discov. 2004;12(11):935-49.
- [10] Meng XY, Zhang HX, Mezei M and Cui M. Molecular docking: a powerful approach for structure-based drug discovery. Curr Comput Aid Drug. 2011;7(2):146-57.
- [11] Pawar DC, Gaikwad SV, Kamble SS, Gavhane PD, Gaikwad MV, Dawane BS. Design, synthesis, docking and biological study of pyrazole-3, 5-diamine derivatives with potent antitubercular activity. Chem Methodol. 2022; 6:677-90.